CLAIMS

What is claimed is:

- 1. A method for recovering and producing C₄-C₆ dicarboxylates from an alkaline waste solution generated in a caprolactam preparation process, comprising the steps of:
 - (1) neutralizing the alkaline waste solution generated from the caprolactam preparation process and adjusting a pH value thereof with sulfuric acid to separate into an aqueous phase and an organic phase; adding nitric acid to the organic phase for oxidizing and converting valuable substances contained in the organic phase into dicarboxylic acids, thereby obtaining an oxidized reaction mixture containing dicarboxylic acids;
 - (2) introducing the oxidized reaction mixture obtained from the step (1) into a two-stage concentration apparatus; in first-stage concentration, distilling out low boiling-point monocarboxylic acids and nitric acid; in second-stage concentration, decomposing remaining nitric acid and nitrocompounds, so as to obtain crude concentrates mainly containing C₄-C₆ dicarboxylic acids;
 - (3) adding C₁-C₄ alkyl alcohol to the crude concentrates mainly containing C₄-C₆ dicarboxylic acids obtained from the step (2), and proceeding two-stage esterification, whereby half-esterified intermediates are obtained in first-stage esterification, and crude dicarboxylates are obtained in second-stage esterification; and
 - (4) distilling the crude dicarboxylates obtained from the step (3), so as to get single-species dicarboxylates or a mixture of dicarboxylates.
- 2. The method of claim 1, wherein in the step (1), at least one oxidant other than nitric acid is added to the organic phase.

- 3. The method of claim 2, wherein the oxidant is selected a group consisting of hydrogen peroxide, perchloric acid and potassium permanganate.
- 4. The method of claim 1, wherein concentration of nitric acid is 10 to 90%.
- 5. The method of claim 1, wherein a weight ratio of nitric acid to the organic phase is 0.5 to 30:1.
- 6. The method of claim 2, wherein the oxidant is added in a ratio of 0 to 5%.
- 7. The method of claim 1, wherein oxidation and conversion of the step (1) are performed under conditions including: at least one reaction stage, reaction temperature of a range from 10 to 150°C, and reaction mixture flowing from first to last reaction stage with the reaction temperature being increased gradually at stage intervals of 5 to 30°C, reaction time for each stage being set as 5 minutes to 4 hours, and reaction pressure being absolute pressure 0.5 to 2kg/cm².
- 8. The method of claim 1, wherein in the step (2), the first-stage concentration is performed at temperature of 50 to 120°C and absolute pressure of 0.2 to 1.5 kg/cm², the second-stage concentration is carried out at temperature of 120 to 200°C and absolute pressure of 0.5 to 2.0 kg/cm².
- 9. The method of claim 1, wherein the C₁-C₄ alkyl alcohol used in the step (3) is selected from a group consisting of methanol, ethanol, propanol and butanol.
- 10. The method of claim 1, wherein a weight ratio of the C₁-C₄ alkyl alcohol to the crude concentrates of dicarboxylic acids is 1-15:1.
- 11. The method of claim 9, wherein a weight ratio of the C₁-C₄ alkyl alcohol to the crude concentrates of dicarboxylic acids is 1-15:1.

- 12. The method of claim 1, wherein a catalyst is added in the step (3), and selected from a group consisting of sulfuric acid, phosphoric acid, nitric acid, alkanesulfonic acids, benzenesulfonic acid and cation exchange resin.
- 13. The method of claim 12, wherein the catalyst is added in an amount of 0 to 5%.
- 14. The method of claim 1, wherein in the step (3), the first-stage esterification is performed at temperature of 40 to 120°C, absolute pressure of 0.2 to 1.2 kg/cm², and esterification time of 0.5 to 8 hours; and the second-stage esterification is carried out at temperature of 80 to 200°C, absolute pressure of 0.8 to 2.5 kg/cm², and esterification time of 0.5 to 8 hours.
- 15. The method of claim 1, wherein a distillation apparatus used in the step (4) is of tray or packing type.
- 16. The method of claim 14, wherein distillation of the step (4) is performed by using 10 to 100 trays theoretically, and under absolute pressure of 0.02 to 1.0 kg/cm² and temperature of 70 to 250°C.
- 17. The method of claim 1, comprising the steps of:
 - (1) neutralizing the alkaline waste solution generated from the caprolactam preparation process and adjusting a pH value thereof with sulfuric acid to separate into an aqueous phase and an organic phase; adding nitric acid of 10 to 90% concentration to the organic phase in a weight ratio of nitric acid to the organic phase at 0.5 to 30:1, and oxidizing and converting valuable substances contained in the organic phase into dicarboxylic acids, thereby obtaining an oxidized reaction mixture containing dicarboxylic acids;
 - (2) introducing the oxidized reaction mixture obtained from the step (1) into a two-stage concentration apparatus, wherein first-stage concentration is

performed at temperature of 50 to 120°C and absolute pressure of 0.2 to 1.5 kg/cm², and second-stage concentration is carried out at temperature of 120 to 200°C and absolute pressure of 0.5 to 2.0 kg/cm², so as to obtain crude concentrates mainly containing C₄-C₆ dicarboxylic acids;

- (3) adding C₁-C₄ alkyl alcohol to the crude concentrates mainly containing C₄-C₆ dicarboxylic acids obtained from the step (2), and proceeding two-stage esterification, wherein first-stage esterification is performed at temperature of 40 to 120°C, absolute pressure of 0.2 to 1.2 kg/cm², and esterification time of 0.5 to 8 hours; and second-stage esterification is carried out at temperature of 80 to 200°C, absolute pressure of 0.8 to 2.5 kg/cm², and esterification time of 0.5 to 8 hours, so as to obtain crude dicarboxylates; and
- (4) introducing the crude dicarboxylates obtained from the step (3) into a tray- or packing-type distillation tower for performing distillation by using 10 to 100 trays theoretically, and under absolute pressure of 0.02 to 1.0 kg/cm² and temperature of 70 to 250°C, so as to get single-species dicarboxylates or a mixture of dicarboxylates.